

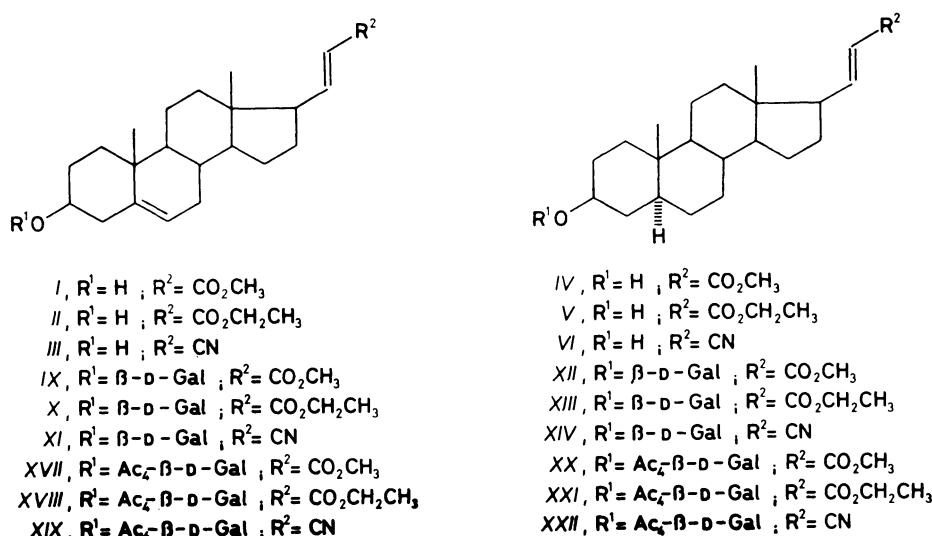
PREPARATION OF 3-(β -D-GALACTOPYRANOSYLOXY)ANDROSTANE DERIVATIVES WITH UNSATURATED SIDE CHAIN IN POSITION 17 β

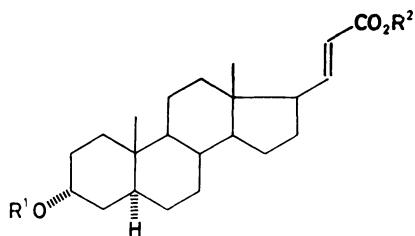
Hana CHODOUNSKÁ and Vladimír POUZAR

*Institute of Organic Chemistry and Biochemistry,
Czechoslovak Academy of Sciences, 166 10 Prague 6*

Received November 13, 1989
Accepted November 15, 1989

The 3-hydroxy derivatives *I*–*VIII* were transformed into the β -D-galactosides *IX*–*XVI* respectively, by the silver silicate promoted¹ glycosylation with 2,3,4,6-tetra-O-acetyl- α -D-galactopyranosyl bromide. Without further purification, the obtained acetylated galactosides (unseparable from by-products) were hydrolyzed to give free β -D-galactosides whose structure was confirmed by acetylation to the corresponding tetraacetates *XVII*–*XXIV*. The ¹H NMR signals correspond to reported data for steroid^{2,3} and sugar⁴ moieties.





VII, R¹=H; R²=CH₃

VIII, R¹=H; R²=CH₂CH₃

XV, R¹=β-D-Gal; R²=CH₃

XVI, R¹=β-D-Gal; R²=CH₂CH₃

XXIII, R¹=Ac₄-β-D-Gal; R²=CH₃

XXIV, R¹=Ac₄-β-D-Gal; R²=CH₂CH₃

β-D-Gal = β-D-galactopyranosyl

Ac₄-β-D-Gal = 2,3,4,6-tetra-O-acetyl-β-D-galactopyranosyl

TABLE I
Yields and physico-chemical data of β-D-galactosides IX—XVI

β-D-Galactoside	Yield, % Solvent ^a	M.p. °C	[α] _D (c) ^b	Formula M.w.	Calculated/Found		
					% C	% H	% N
IX	61 A + E	224—226 (2·2)	—37° (2·2)	C ₂₉ H ₄₄ O ₈ 520·7	66·90 67·18	8·52 8·77	— —
X	45 A + E + C	234—237 (2·4)	—36° (2·4)	C ₃₀ H ₄₆ O ₈ 534·7	67·39 67·25	8·67 8·55	— —
XI	73 C + M	270—274 (0·6)	—32° (0·6)	C ₂₈ H ₄₁ NO ₆ 487·6	68·97 68·88	8·47 8·33	2·87 3·01
XII	65 C + A + M	214—217 (2·5)	+6° (2·5)	C ₂₉ H ₄₆ O ₈ 522·7	66·64 66·43	8·87 9·01	— —
XIII	68 A + E + C	226—230 (3·1)	+8° (3·1)	C ₃₀ H ₄₈ O ₈ 536·7	67·14 66·98	9·01 8·75	— —
XIV	57 C + M	255—275 decomp.	+14° (0·8)	C ₂₈ H ₄₃ NO ₆ 489·7	68·68 68·76	8·85 8·99	2·86 3·01
XV	75 A + E	214—218 (2·5)	+27° (2·5)	C ₂₉ H ₄₆ O ₈ 522·7	66·64 66·42	8·87 8·62	— —
XVI	75 T + A	214—216 (3·1)	+32° (3·1)	C ₃₀ H ₄₈ O ₈ 536·7	67·14 66·95	9·01 8·88	— —

^a Solvents: A ether, E ethanol, C chloroform, M methanol, T toluene; ^b measured in chloroform.

EXPERIMENTAL

Melting points were determined on a micro melting point apparatus Boetius (G.D.R.). Optical rotations were measured at 25°C on a Perkin-Elmer 141 MC polarimeter. Infrared spectra were recorded on Perkin-Elmer PE 580 spectrometer. ^1H NMR spectra were measured on spectrometer Varian XL-200 (FT mode, 200 MHz) in deuteriochloroform with tetramethylsilane as internal standard. Chemical shifts are given in ppm (δ -scale), coupling constants (J) and widths of multiplets (W) are given in Hz. All parameters were obtained by first-order analyses. Column chromatography was performed on Silpearl (Kavalier, Votice), thin-layer chromatography on silica gel G according to Stahl (ICN Biochemicals). Prior to evaporation, solutions in organic solvents were dried over anhydrous sodium sulfate. Solvents were evaporated in vacuo (about 2 kPa). Analytical samples were dried over phosphorus pentoxide at 40°C/26 Pa for 12 h.

TABLE II
IR spectra ($\tilde{\nu}$, cm^{-1} ; KBr pellet) of β -D-galactosides *IX*–*XVI*

Compound	O-H	C≡N	C=O	C=C	C-O
<i>IX</i>	3 450	—	1 718	1 652	1 065
<i>X</i>	3 510 sh, 3 420, 3 330 sh, 3 110	—	1 718	1 653	1 092, 1 077, 1 047
<i>XI</i>	3 485, 3 420, 3 350, 3 120	2 225	—	1 631	1 153, 1 095, 1 077, 1 046
<i>XII</i>	3 540 sh, 3 420, 3 340 sh, 3 130 sh	—	1 724	1 653	1 093, 1 074, 1 044
<i>XIII</i>	3 500 sh, 3 415, 3 330 sh, 3 120 sh	--	1 713	1 656	1 093, 1 075, 1 044
<i>XIV</i>	3 520, 3 415, 3 350 sh, 3 120	2 225	—	1 631	1 148, 1 094, 1 078, 1045
<i>XV</i>	3 475, 3 400, 3 160	—	1 713	1 649	1 104, 1 058, 1 040 sh
<i>XVI</i>	3 410	—	1 703	1 652	1 085 sh, 1 067, 1 040

General Procedure for Preparation of Galactosides *IX—XVI*

A dry mixture of 3-hydroxy derivative^{2,3} (1 mmol), silver silicate¹ (1·4 g) and ground molecular sieve 4A (2·0 g) was stirred in vacuo (10 Pa) for 4 h. The flask was then filled with argon under slight overpressure (about 5 kPa) and 1,2-dichloroethane (20 ml) was injected through a septum. The mixture was stirred at room temperature for 20 min and a solution of 2,3,4,6-tetra-O-acetyl- α -D-galactopyranosyl bromide (500 mg, 1·2 mmol) in 1,2-dichloroethane (3 ml) was added (through septum). After stirring at room temperature for 20 h, the catalyst was removed by filtration through a column of Celite. The column was washed with chloroform-ether (4 : 1) and combined filtrates were washed with 5% aqueous sodium hydrogen carbonate and water.

TABLE III
Yields and physico-chemical data of β -D-galactosides tetraacetates *XVII—XXIV*

Acetate	Yield, % Solvent ^a	M.p. °C	$[\alpha]_D$ (c) ^b	Formula M.w.	Calculated/Found		
					% C	% H	% N
<i>XVII</i>	65 A + E	222—223	—14° (5·0)	$C_{37}H_{52}O_{12}$ 688·8	64·52 64·81	7·61 7·75	— —
<i>XVIII</i>	61 A + C	204—205	—15° (2·3)	$C_{38}H_{54}O_{12}$ 702·8	64·94 65·10	7·74 7·89	— —
<i>XIX</i>	71 A + P	161—162	—11° (1·7)	$C_{36}H_{49}NO_{10}$ 655·8	65·94 66·12	7·53 7·66	2·14 2·11
<i>XX</i>	76 A + C	231—233	+18° (3·9)	$C_{37}H_{54}O_{12}$ 690·8	64·33 64·12	7·88 7·98	— —
<i>XXI</i>	85 A	219—222	+17° (1·9)	$C_{38}H_{56}O_{12}$ 704·9	64·75 65·02	8·01 7·87	— —
<i>XXII</i>	72 E + P	164—165	+22° (1·7)	$C_{36}H_{51}NO_{10}$ 657·8	65·73 65·89	7·81 7·66	2·13 2·01
<i>XXIII</i>	83 A + P	162—164	+4° (2·8)	$C_{37}H_{54}O_{12}$ 690·8	64·33 64·31	7·88 8·02	— —
<i>XXIV</i>	48 A + P	161—162	+4° (1·4)	$C_{38}H_{56}O_{12}$ 704·9	64·75 64·88	8·01 8·22	— —

^a Solvents: A ether, E ethyl acetate, C chloroform, P light petroleum; ^b measured in chloroform

The solvents were evaporated and the residue was dissolved in methanol (30 ml; in the case of methyl esters *IX*, *XII*, and *XV* and nitriles *XI*, and *XIV*) or in ethanol (30 ml; in the case of ethyl esters *X*, *XIII*, and *XVI*). After addition of 20 drops of 5% solution of the corresponding sodium alkoxide, the mixture was stirred at room temperature for 3 h, neutralized with dry ice (about 300 mg) and the solvent was evaporated in vacuo. The residue was chromatographed on a column of Silpearl (40 g) in chloroform-methanol (9 : 1, for compounds *IX*, *XI*, *XII*, *XIV*, and *XV*) or in chloroform-ethanol (9 : 1, for compounds *X*, *XIII*, and *XVI*). The product was further purified by crystallization. Solvents used for crystallization, yields, melting points, optical rotations and elemental analyses are given in Table I. Infrared spectra are given in Table II.

General Procedure for Preparation of Tetraacetates *XVII*—*XXIV*

A solution of the galactoside (0.2 mmol) in a mixture of pyridine (1 ml) and acetic anhydride (1 ml) was allowed to stand overnight at room temperature. The mixture was coevaporated with toluene (3×), the residue was dissolved in ether, filtered through Celite and the solvent was evaporated in vacuo. The product was further purified by crystallization. Solvents used for crystallization, yields, melting points, optical rotations and elemental analyses are given in Table III. Infrared and ^1H NMR spectra are given in Tables IV and V, respectively.

Our thanks are due Dr S. Vašičková and Dr M. Buděšínský for measurement and interpretation of IR and ^1H NMR spectra. Elemental analyses were carried out in the Analytical Laboratory of this Institute (Dr V. Pechanec, Head).

TABLE IV

IR spectra ($\tilde{\nu}$, cm^{-1} ; tetrachloromethane) of β -D-galactosides tetraacetates *XVII*—*XXIV*

Compound	$\text{C}\equiv\text{N}$	$\text{C}=\text{O}^a$	$\text{C}=\text{O}^b$	$\text{C}=\text{C}$	$\text{C}-\text{O}$
<i>XVII</i>	—	1 758	1 726	1 652	1 221, 1 058
<i>XVIII</i>	—	1 757	1 720	1 652	1 220, 1 055
<i>XIX</i>	2 225	1 757	—	1 631	1 222
<i>XX^c</i>	—	1 750	1 715	1 650	
<i>XXI</i>	—	1 758	1 719	1 652	1 220
<i>XXII</i>	2 225	1 757	—	1 631	1 222
<i>XXIII</i>	—	1 758	1 726	1 652	1 220
<i>XXIV^c</i>	—	1 750	1 707	1 650	

^a Carbonyl of galactoside acetates; ^b carbonyl of unsaturated ester; ^c in chloroform solution.

TABLE V
¹H NMR spectral parameters (in deuteriochloroform) for galactoside tetraacetates XVII–XXIV^a

Parameter	XVII ^b	XVIII ^c	XIX	XX ^b	XXI ^c	XXII	XXIII ^b	XXIV ^b
Steroid unit								
H-18 (s, 3 H)	0.66	0.66	0.63	0.63	0.63	0.62	0.62	0.62
H-19 (s, 3 H)	0.99	1.00	1.00	0.79	0.79	0.78	0.78	0.78
H-21 (dd, 1 H)	5.78	5.78	5.28	5.77	5.77	5.26	5.78	5.77
H-20 (dd, 1 H)	7.01	6.94	6.69	6.94	6.93	6.68	6.95	6.94
H-3 (m, 1 H)	3.49 ^d	3.49 ^d	3.36 ^d	3.55 ^d	3.56 ^d	3.55 ^d	3.92 ^e	3.92 ^e
H-6 (m, 1 H)	5.35	5.35	5.36	5	5	5	f	f
J(20, 21)	15.6	16.4	15.6	15.6	16.4	15.6	15.6	15.6
J(17, 20)	8.0	8.0	8.1	8.0	8.0	8.2	8.0	8.0
J(17, 21)	1.1	1.1	1.2	1.2	1.1	1.2	1.1	1.1

	Sugar unit					
H-1 (d, 1 H)	4.55	4.54	4.55	4.54	4.49	4.49
H-2 (dd, 1 H)	5.19	5.18	5.17	5.18	5.17	5.22
H-3 (dd, 1 H)	5.02	5.01	5.01	5.00	5.00	5.02
H-4 (dd, 1 H)	5.38	5.38	5.38	5.37	5.37	5.38
H-5 (dt, 1 H)	3.89	3.89	3.88	3.88	3.88	3.87
H-6a (dd, 1 H)	4.10	4.10	4.10	4.12	4.09	4.10
H-6b (dd, 1 H)	4.19	4.19	4.19	4.20	4.18	4.17
J(1, 2)	7.7	7.7	7.8	7.7	7.7	7.7
J(2, 3)	10.5	10.5	10.6	10.6	10.5	10.6
J(3, 4)	3.4	3.4	3.4	3.4	3.4	3.4
J(4, 5)	1.1	1.1	1.2	1.1	1.0	1.1
J(5, 6a)	6.8	6.7	6.7	6.9	6.8	6.7
J(5, 6b)	6.8	7.0	7.0	6.6	6.8	6.8
J(6a, 6b)	11.1	11.0	11.3	11.1	11.1	11.1
OAc (s, 3 H)	2.14	2.15	2.14	2.14	2.14	2.15
OAc (s, 3 H)	2.06	2.06	2.06	2.05	2.05	2.05
OAc (s, 3 H)	2.04	2.04	2.04	2.04	2.04	2.04
OAc (s, 3 H)	1.98	1.98	1.98	1.98	1.98	2.00

^a For other conditions see Experimental; ^b other signal 3.72 s, 3 H (COOCH₃); ^c other signals 4.18 q, 2 H and 1.28 t, 3 H (COOCH₂CH₃, J = 7.2); ^d W = 32; ^e W = 16; ^f undeterminable value.

REFERENCES

1. Paulsen H., Kutschker W.: *Carbohydr. Res.* **120**, 25 (1983).
2. Pouzar V., Chodounská H., Sameš D., Drašar P., Havel M.: *Collect. Czech. Chem. Commun.* **55**, 1243 (1990).
3. Pouzar V., Chodounská H., Drašar P., Havel M.: *Collect. Czech. Chem. Commun.* submitted.
4. Jimeno M. L., Martin-Lomas M., Alemany A.: *Magn. Reson. Chem.* **23**, 1082 (1985).

Translated by the author (V.P.).